#### UG-III/Chem-X(H)/PR/20

# 2020

## **CHEMISTRY**

[HONOURS]

Paper : X

(PRACTICAL)

Full Marks: 80

Time:

The figures in the right-hand margin indicate marks.

#### **GROUP-A**

(Inorganic)

(Marks: 40)

Answer any **four** from the following:

 $10 \times 4 = 40$ 

1. a) Define solubility and solubility product. Explain common ion effect with example.

2+2+3

- b) What do you mean by semi-micro qualitative inorganic analysis? 2
- c) Give the formula of sodium cobaltinitrite. 1
- 2. a) In case of separation of metal ions through qualitative analysis, mention the number of groups, corresponding metal ions assigned in the respective groups and specify the corresponding group reagents. 2+2+2
  - b) Explain why Pb<sup>2+</sup> is included in both Gr-I and Gr-II?
  - c) How  $Mg^{2+}$  is detected?

2

[Turn over]

- 3. a) What is flame test? Why conc. HCl is used in the flame test?
  - b) Write down the reactions involved in the oxidative fusion test for  $Cr^{3+}$ .
  - c) You are given a sample of CoO, how could you identify Co<sup>2+</sup> through Borax bead test? Give the reactions starting from heating the borax.

4

4. a) Give the formula of ammonium molybdate. How phosphate is detected? Give reactions.

2+3

- b) Draw the structure of DMG? Write down the balanced equation of the reaction of Ni<sup>2+</sup> with DMG in ammoniacal medium. 2+3
- 5. a) Give the tests in tabular form for the cation and anion in  $PbSO_4$  as an insoluble residue.

5

- b) How chloride can be detected in presence of bromide. Present it in tabular from. 5
- 6. Write short notes on (any **two**): 5+5
  - a) Chromyl chloride test
  - b) Brown ring test
  - c) Phosphate separation

#### **GROUP-B**

### [Organic and Industrial]

[Marks : 40]

- 7. Answer any **one** from the following:  $20 \times 1 = 20$ 
  - A. a) Discuss the principle of estimation of aniline with proper equation. 5
    - b) Discuss the procedure of estimation of aniline. 5
    - c) What is starch? What is the composition of starch? 5
    - d) How will you calculate the equivalent weight of KBrO<sub>3</sub>? What is the functional group present in aniline? Is it involved directly in the estimation of aniline?

2+1+2=5

- B. a) Discuss the principle of estimation of glucose with proper equation. 5
  - b) Discuss the procedure of estimation of glucose. 5
  - c) What are Fehling A and Fehling B solutions?
  - d) 'Reaction between glucose and Fehling solution is a redox process.' Comment on this statement with proper reasons.

5

- 8. Answer any **one** from the following:  $20 \times 1 = 20$ 
  - A. a) What is the cause of hardness of water?
    How is the hardness of water expressed?
    Can EDTA be used as primary standard?
    2+1+2=5
    - b) Discuss the theory for determination of total hardness of water. 5
    - c) Discuss the titration procedures for the measurement of total hardness of water.
    - d) What is EBT? What kind of indicator is it? What is buffer solution? 2+1+2=5
  - B. a) What is pyrolusite? What do you mean by available oxygen in pyrolusite? In permanganometry no indicator is used -why? 1+2+2=5
    - b) Discuss the theory for the determination of available oxygen in pyrolusite. 5
    - c) Discuss the titration procedure for the estimation of available oxygen in pyrolusite.
    - d) Calculate equivalent weight of KMnO<sub>4</sub> solution in acid medium. Is KMnO<sub>4</sub> solution primary standard? Explain.

5

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125(Sc)/PR [3] [*Turn over*]

125(Sc)/PR

[4]